



University of Nevada
Cooperative Extension

Special Publication 12-04



Boron- and salt-tolerant trees and shrubs for northern Nevada

Heidi Kratsch

Western Area Horticulture Specialist

University of Nevada Cooperative Extension

What is boron?

Boron is a mineral that, in small quantities, is essential for plant growth and development, but becomes toxic at levels above 0.5 to 1 part per million (ppm) in the soil. Excess boron may be naturally present in the soil, and it can accumulate by irrigating with water high in boron. Boron occurs naturally in arid soils originating from geologically young deposits. It may accumulate in soils that are heavily amended with borate-containing fertilizers. In Nevada, well water or water from springs near geothermal areas or earthquake faults may contain high concentrations of boron. Excessive concentrations are rarely found in surface water. Boron can also enter the soil from sewage effluent or from irrigation with reclaimed water. Boron is mobile in soils, and will move up in the soil horizon as moisture evaporates and down when soils are irrigated with fresh water.

Boron and salty soils

Soils high in boron may also be salty. Soils become salty for some of the same reasons as they become high in boron. In cold climates, road salt may also contribute to the saltiness of landscape soils. Salty soils contain higher than normal levels of mineral salts, such as chlorides, sulfates, ammonium or even sodium. High levels of mineral salts in soils damage sensitive plants. Alone, these mineral elements are called *specific ions*, and plants can be sensitive to individual specific ions in addition to the overall soil salinity. Boron is a specific ion, but it is toxic at such low levels that it must be tested for separately from soil salinity. Only a few landscape plants are known to be tolerant of both high soil salt and the presence of high levels of boron (Table 1).

Symptoms of boron toxicity

Symptoms of boron toxicity occur on older leaves of plants, and look much like symptoms of high salinity, or of iron or manganese deficiencies. Early stages of boron toxicity show up as chlorosis (yellowing) of leaf tips, and are followed by necrosis (death) of leaf margins and of the tissues in between leaf veins. In particular, later stages of boron toxicity exhibit as blackened areas or irregularly shaped black spots along leaf margins or between leaf veins, depending on the plant species affected (page 4). Conifer needles die from the tip downward, with the most extensive damage occurring on older needles. It must be emphasized that nutrient deficiencies or specific ion toxicities cannot be accurately diagnosed by observing symptoms. Soil, water and/or tissue samples should be tested where a definitive diagnosis is required.

Managing plants in boron-laden soils

The most important step in managing high-boron landscape soils is proper plant selection. Do not install plants that are known to be sensitive to boron (Table 2). The boron tolerance of many landscape plants is unknown, and developers may encounter limitations in local availability of many boron-tolerant plant varieties due to their lack of cold-hardiness. The historical approach has been to irrigate these soils more frequently with fresh water to leach salts and boron below the plant root zone. However, some boron-infested areas have high water tables, high pH and/ or may have soils with poor structural characteristics, such as sodium-affected (sodic) soils, leading to insufficient drainage. These conditions exacerbate boron toxicity, and irrigating more frequently leads to overwatering. Another confounding factor is the presence of hardpan in the soil horizon, which also interferes with soil water drainage. A more recent approach to planting in boron-laden landscapes is to install plants above the native soil. This can be achieved by bringing in higher quality topsoil or other appropriate planting substrate and creating "planting mounds" for trees and shrubs. To be effective, mounds must be at least 12 inches in depth and extend beyond the drip line of landscape plants.

This practice lifts most of the plant feeder roots out of the boron-laden zone and improves aeration and drainage. It is most effective when plants are clustered in groups on a mounded island, rather than individually mounded in a linear design.

Will adding soil amendments help?

Gypsum (calcium sulfate) is often recommended to improve soil structure, change soil pH and/or leach salts from soils. Gypsum, used at the proper rate, will leach sodium ions from sodic soils. Sodic soils are high in exchangeable sodium relative to calcium and magnesium. High levels of sodium in soils leads to breakdown in soil structure and results in fine-textured soils that lack aeration or good water infiltration. The calcium in gypsum used to remedy such soils displaces sodium ions, which must be leached below plant root systems using good-quality water. However, the sodium must have some place to go, and leaching of sodic soils is ineffective in areas with drainage problems, such as floodplains or areas with hardpan layers in the soil horizon. Gypsum is required in large quantities, takes a long time to have an effect and the results are temporary. So leaching with gypsum must be repeated regularly, which is impractical in urban landscapes. Studies show that use of gypsum to lower soil sodium is most effective in agricultural areas, where regular disturbance of the soil is acceptable. Gypsum does not affect the level of other mineral salts, including boron, in soils. Gypsum can improve the structure of heavy clay soils, but does not improve water-holding capacity or fertility of sandy soils. Finally, gypsum does not significantly affect the soil pH.

Sulfur is often recommended for lowering the pH of alkaline soils. It is effective for use prior to landscape establishment, but it takes a long time (sometimes years) to have an effect on pH. It can cause damage to plants and increase soil salinity if used regularly on established landscapes. The pH of a soil refers to its acidity or alkalinity. Plants grow best at a pH between 5.5 and 7.5. Alkaline soils have a pH above 7.5, which causes some soil nutrients to become unavailable to plant roots. Obviously, lowering the pH of our soils will be beneficial to plant growth and health. Unfortunately, many of the soils in northern Nevada have a high level of “free lime” or calcium carbonate. This means that the soil is buffered and resists significant changes in pH. Adding sulfur to such soils is a costly waste of time and will not be effective. Elemental sulfur may be used in localized areas around plant roots to alleviate pH-related nutrient deficiencies, and it may also be used to leach sodium from sodic soils. It has no effect on soil structure or other mineral salts, including boron.

Organic matter is an easy, safe and effective amendment that improves the structure and water-holding capacity of sandy soils and, over time, loosens compacted or heavy clay soils. Organic mulches moderate soil temperatures around plant roots, reduce germination of weed seeds by blocking light and break down slowly, gradually lowering soil pH and providing nutrients for landscape plants. Organic mulches include aged bark or wood chips, pesticide- and disease-free grass clippings and decomposing leaf litter. You can rarely go wrong by adding organic matter to your soil, and it is safe to use in established landscapes. The only caveat is that homeowners and landscapers must use caution when applying organic mulches in fire-prone areas because most organic mulches are highly combustible. In these areas, use of organic mulch should be limited to the space within the drip line of trees and not connected to other mulched areas. Well-rotted animal manure is another effective organic amendment but its overuse may add to the saltiness of already salty soils. Finally, even organic matter will not significantly lower soil salts, including boron.

Symptoms of boron toxicity are species-dependent. These trees are growing in the same landscape but show varying symptoms depending on their tolerance for high soil salts, pH and boron.



Boron toxicity symptoms on bur oak present as browning at the tips of the foliage. Many oaks are intolerant of excess boron in the soil.

The needles of blue spruce in boron-laden soils brown from the tip down. Older needles are affected first, and over time, the tree will lose more needles than it can replace.



Sweetgum is moderately tolerant of salty and boron-laden soils, but it is very sensitive to high pH. This tree has severe iron-deficiency chlorosis as a result of the high pH of the soil in this landscape.

Table 1. Boron- and Salt-Tolerant Tree and Shrub List

TREES

Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Acer ginnala</i> (Amur maple)	3—8	Moderate	Moderate	Low
<i>Calocedrus decurrens</i> (Incense-cedar)	5—8	Moderate	Moderate	Low
<i>Carpinus betulus</i> (European hornbeam)	4—8	Moderate	Low	Medium
<i>Cedrus atlantica</i> (Atlas cedar)	6—9	High	Low	Medium
<i>Cedrus deodara</i> (Deodar cedar)	7—9	High	Low	Medium
<i>Chionanthus virginicus</i> (White fringetree)	4—9	Moderate	Moderate	Medium
<i>Crataegus phaenopyrum</i> (Washington hawthorn)	3—8	High	Low to Moderate	Low
<i>Cupressus arizonica</i> Arizona cypress	7—9	Moderate	Low to Moderate	Low
<i>Cupressus sempervirens</i> (Italian cypress)	7—9	High	Moderate	Low
<i>Eucalyptus polyanthemos</i> (Silver dollar gum)	7	Moderate to High	High	Low
<i>Forestiera neomexicana</i> (New Mexico privet)	5—9	Moderate	Moderate	Low
<i>Fraxinus oxycarpa</i> 'Raywood' (Raywood Ash)	5—8	High	Low	Low
<i>Fraxinus pennsylvanica</i> (Green ash)	3—9	Moderate	Moderate	Low
<i>Fraxinus velutina</i> (Arizona ash)	(6)7—10	High	Moderate	Low
<i>Gymnocladus dioicus</i> (Kentucky coffeetree)	3—8	Moderate	Moderate	Low
<i>Juglans regia</i> (English walnut)	4—9	Moderate	Low	Medium
<i>Liquidambar styraciflua</i> (Sweetgum) NOT 'Festival'	5—9	Moderate	Moderate	Low

Table 1. Boron- and Salt-Tolerant Tree and Shrub List (cont.)

TREES

Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Liriodendron tulipifera</i> (Tulip tree)	4—9	Moderate	Low	Medium
<i>Malus hybrids</i> (Crabapple)	3—9	High	Moderate to High	Medium
<i>Morus alba</i> (White mulberry) NOT fruitless	5—9	Moderate	High	Low
<i>Pinus bungeana</i> (Lacebark pine)	(4)5—7	Moderate	Moderate	Medium
<i>Pinus thunbergii</i> (Japanese black pine)	(5)6—8	Moderate	Moderate	Medium
<i>Platanus x acerifolia</i> (London planetree)	4—9	Moderate	High	Low
<i>Populus fremontii</i> (Western cottonwood)	Zone 3—8	High	Moderate	Medium
<i>Prunus cerasifera</i> 'Atropurpurea' (Purple-leaf plum)	4—8	High	Moderate	Medium
<i>Prunus serrulata</i> (Flowering cherry)	5—9	Moderate	Moderate	Medium
<i>Salix babylonica</i> (Weeping willow)	(5)6—8	High	Moderate	Medium
<i>Salix matsudana</i> 'Tortuosa' (Corkscrew willow)	4—7	High	Moderate to High	Medium
<i>Sophora japonica</i> (Japanese pagoda tree)	4—8	Moderate	Moderate	Medium
<i>Ulmus parvifolia</i> (Lacebark elm)	5—9	Moderate	Moderate	Low
<i>Ulmus pumila*</i> (Siberian elm)	4—9	Moderate	High	Low
<i>Zelkova serrata</i> (Japanese zelkova)	5—8	Moderate	Moderate	Low

* May be planted only in unincorporated areas in Washoe County.

Table 1. Boron- and Salt-Tolerant Tree and Shrub List (cont.)

SHRUBS

Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Acer campestre</i> (Hedge maple)	(4)5—8	Moderate	Moderate	Medium
<i>Amelanchier spp.</i> (Serviceberry)	3—6	Moderate	Moderate	Low—Medium
<i>Aronia melanocarpa</i> (Black chokeberry)	3—8(9)	Moderate	Moderate	Medium
<i>Atriplex spp.</i> (Saltbush, Shadscale)	3—10	Moderate	Moderate	Very Low
<i>Buxus microphylla</i> var. <i>japonica</i> (Japanese box-wood)	6—9	High	Moderate	Medium
<i>Ceanothus spp.</i> (Wild lilac)	4—8	High	Unknown	Low—Medium
<i>Juniperus chinensis</i> 'Kaizuka' Hollywood juniper	4—9	High	Moderate	Low
<i>Juniperus chinensis</i> 'Pfitzerana' Pfitzer juniper	4—8	High	Moderate	Low
<i>Teucrium chamaedrys</i> (Wall Germander)	5—9	Moderate	Moderate	Low
<i>Cercocarpus betuloides</i> (Alderleaf mountain mahogany)	3—6	Moderate	Moderate	Low
<i>Cercocarpus ledifolius</i> Littleleaf mountain mahogany	3—8	Moderate	Moderate	Low
<i>Chaenomeles speciosa</i> (Flowering quince)	4—8(9)	Moderate	Moderate	Medium
<i>Potentilla fruticosa</i> (Shrubby cinquefoil)	2—7	Moderate	Moderate	Medium
<i>Prunus besseyi</i> (Western sand cherry)	3—6	Moderate	Moderate	Low

Table 1. Boron- and Salt-Tolerant Tree and Shrub List (cont.)

SHRUBS

Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Rhus aromatic</i> (Fragrant sumac)	3—9	Moderate	Moderate	Low
<i>Rhus trilobata</i> (Threeleaf sumac)	4—7	Moderate	Moderate	Low
<i>Salix purpurea</i> ‘Nana’ (Basket willow)	3—7	Moderate	Moderate	Medium
<i>Shepherdia argentea</i> (Silver buffaloberry)	3—6	Moderate	Moderate	Medium

Table 2. Trees and shrubs known to be sensitive to boron

Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Acer saccharinum</i> (Silver maple)	3—9	Low	Low	Medium
<i>Aesculus x carnea</i> (Red horsechestnut)	4—7	Low	Low	Medium
<i>Alnus rhombifolia</i> (White alder)	5—8	Low	Low	Medium
<i>Betula pendula</i> (European white birch)	2—7	Low	Low	Medium
<i>Carya illinoensis</i> (Pecan)	5—9	Low	Low	Medium
<i>Catalpa spp.</i> (Catalpa)	Vary by species	Low	Low	Medium
<i>Cercis spp.</i> (Redbud)	Vary by species	Low	Low to Moderate	Medium
<i>Gleditsia triacanthos</i> var. <i>inermis</i> (Thornless honeylocust)	3—9	Low	Moderate	Medium

Table 2. Trees and shrubs known to be sensitive to boron (cont.)

TREES				
Species	USDA Hardiness Zone	Boron Tolerance	Salt Tolerance	Water Use Rating
<i>Juglans nigra</i> (Black walnut)	4—9	Low	Low	Medium
<i>Liquidambar styraciflua</i> 'Festival' (Festival sweetgum)	5—9	Low	Moderate	Low
<i>Magnolia spp.</i> (Magnolia)	Vary by species	Low	Low to Moderate	Medium
<i>Morus alba</i> 'Fruitless' (Fruitless mulberry)	(4)5—9	Low	Moderate	Low
<i>Prunus avium</i> (Sweet cherry)	3—8	Low	Low	Medium
<i>Prunus x domestica</i> (Italian prune)	4—9	Low	Low	Medium
<i>Prunus persica</i> (Peach)	5—9	Low	Low	Medium
<i>Sequoia sempervirens</i> (Coast redwood)	7—9	Low	Unknown	Medium
<i>Ulmus americana</i> (American elm)	2—9	Low	Low	Medium
SHRUBS				
<i>Elaeagnus pungens</i> (Silverberry)	6—9	Low	Moderate	Low
<i>Euonymus japonicus</i> (Japanese euonymus)	(6)7—9	Low	High	Medium
<i>Euonymus japonicus</i> 'Grandifolius' (Bigleaf euonymus)	(6)7—9	Low	Moderate to High	Medium
<i>Juniperus communis</i> 'Hibernica' (Irish juniper)	2—6(7)	Low	Moderate	Low
<i>Mahonia aquifolium</i> (Oregon grape)	(4)5—7	Low	Low	Medium
<i>Photinia x fraseri</i> (Fraser photinia)	7—9	Low	Low	Medium

Table 2. Trees and shrubs known to be sensitive to boron (cont.)

SHRUBS				
Species	USDA Hardiness Zone	Boron Tolerance	Salt Toler- ance	Water Use Rating
<i>Photinia serrulata</i> (Chinese photinia)	6—9	Low	Low	Medium

Acknowledgments

Thanks to UNCE Master Gardener Carol Ort for compiling the list of plants for “Boron Sensitivity in Plants of Northern Nevada.” I thank Jana Vanderhaar, landscape architect with Verdant Connections, for her advice on tree and shrub selection for boron-infested areas. Also, thanks to Steve Churchillo, Reno city forester, for his advice on mound-planting for trees in boron-infested areas.

References

- Carlos, B. 2000. Effects of Boron on Plants. University of Nevada Cooperative Extension.
- Costello, L.R., E.J. Perry, N.P. Matheny, J.M. Henry and P.M. Geisel. 2003. Abiotic Disorders of Landscape Plants. University of California Agriculture and Natural Resources Publication 3420.
- Donaldson, S. 1998. Boron/Plant Interactions. Report compiled for the Steamboat Creek Restoration Steering Committee.
- Kratsch, H.A. 2011. Water-Efficient Landscaping in the Intermountain West. Utah State University Press, Logan, UT.
- Kratsch, H., S. Olsen, L. Rupp, G. Cardon and R. Heflebower. 2008. Soil Salinity and Ornamental Plant Selection. Utah State University Extension Publication HG/Landscaping/2008-02pr.
- Kuhns, M. 2009. TreeBrowser.org. Utah State University Extension. Accessed 6 November 2011.
- Miller, W. W. 2011. personal communication. University of Nevada Reno.
- Wu, L. and L. Dodge. 2005. Landscape Plant Salt Tolerance Selection Guide for Recycled Water Irrigation. A Special Report for the Elvenia J. Slosson Endowment Fund. University of California Agriculture and Natural Resources.

The University of Nevada, Reno is an Equal Opportunity/ Affirmative Action employer and does not discriminate on the basis of race, color, religion, sex, age, creed, national origin, veteran status, physical or mental disability, or sexual orientation in any program or activity it conducts. The University of Nevada employs only United States citizens and aliens lawfully authorized to work in the United States.

Copyright © 2012 University of Nevada Cooperative Extension